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## UCLA chemists create synthetic 'gene-like' crystals for carbon dioxide capture

By Stuart Wolpert | February 11, 2010



Image of 3-D, synthetic DNA-like crystals created by Yaghi, Deng and colleagues

UCLA chemists report creating a synthetic "gene" that could capture heat-trapping carbon dioxide emissions, which contribute to global warming, rising sea levels and the increased acidity of oceans.

The research appears in the Feb. 12 issue of the journal *Science*.

"We created three-dimensional, synthetic DNA-like crystals," said UCLA chemistry and biochemistry professor Omar M. Yaghi, who is a member of the California NanoSystems Institute (CNSI) at UCLA and the UCLA–Department of Energy Institute of Genomics and Proteomics. "We have taken organic and inorganic units and combined them into a synthetic crystal which codes information in a DNA-like manner. It is by no means as sophisticated as DNA, but it is certainly new in chemistry and materials science."

The discovery could lead to cleaner energy, including technology that factories and cars can use to capture carbon dioxide before it reaches the atmosphere.

"What we think this will be important for is potentially getting to a viable carbon dioxide-capture material with ultra-high selectivity," said Yaghi, who holds UCLA's Irving and Jean Stone Chair in Physical Sciences and is director of the CNSI's Center

for Reticular Chemistry. "I am optimistic that is within our reach. Potentially, we could create a material that can convert carbon dioxide into a fuel, or a material that can separate carbon dioxide with greater efficiency."

The research was federally funded by the U.S. Department of Energy's Office of Basic Energy Sciences. The lead author is Hexiang "DJ" Deng, a UCLA graduate student of chemistry and biochemistry who works in Yaghi's laboratory.

"DNA is a beautiful molecule that has a way to code for information," Yaghi said. "How do you code information in a crystal in the same way that DNA does? DJ and I figured out a way to do this. The sequence of organic functionalities that decorates the pores of the crystals is most certainly a unique code."

"DJ has illustrated that one member of a series of materials he has made has 400 percent better performance in carbon dioxide capture than one that does not have the same code," he said.

In the early 1990s, Yaghi invented a class of materials called metal-organic frameworks (MOFs), sometimes described as crystal sponges, in which he can change the components nearly at will. MOFs have pores — openings on the nanoscale in which Yaghi and his colleagues can store gases that are usually difficult to store and transport. Molecules can go in and out of the pores unobstructed. Yaghi and his research team have made thousands of MOFs.

"We have created crystals of metal-organic frameworks in which the sequence of multiple functionalities of varying kind and ratios acts as a synthetic 'gene,'" Yaghi said. "With these multivariate MOFs, we have figured out a way to incorporate controlled complexity, which biology operates on, in a synthetic crystal — taking synthetic crystals to a new level of performance."

"This can be a boon for energy-related and other industrial applications, such as conversion of gases and liquids like carbon dioxide to fuel, or water to hydrogen, among many others," he said.

Yaghi has been collaborating with his former UCLA chemistry colleague and former CNSI director Sir J. Fraser Stoddart on how to take concepts from biology and incorporate them into a synthetic material.

"We hope the materials we are creating will introduce a new class of structures that have controlled complexity," Yaghi said. "Chemists and materials scientists are now able to ask new questions we have never asked before. Also, new tools for characterizing the sequences and deciphering the codes within the crystals will have to be developed."

Carbon dioxide is polluting Earth's atmosphere and damaging coral reefs and marine life — impacts that are irreversible in our lifetime, Yaghi said.

Co-authors on the study are Christian Doonan and Hiroyasu Furukawa, UCLA postdoctoral scholars in Yaghi's laboratory; Ricardo Ferreira, a UCLA visiting undergraduate; John Towne, a former UCLA undergraduate; Carolyn Knobler, a research associate in Yaghi's laboratory; and Bo Wang, a UCLA postdoctoral scholar in Yaghi's laboratory.

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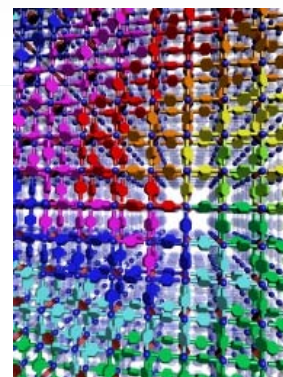


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A method to capture carbon dioxide



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A few years ago, Yaghi spoke at Shanghai's Fudan University, which is known for having one of the best chemistry departments in China. There, he met Deng, who at the time was an undergraduate student at the university. Deng and his colleagues had tried unsuccessfully to make new MOFs.

"DJ told me, 'Professor, we tried a slight variation to make new MOFs and it did not work,'" Yaghi recalled. "I asked, 'How many times did you try?' He said, 'Two or three times.' I said, 'How about 20 times, 30 times? How about 100 times? If it were that easy, why would it need a smart person like you to do it? Success and excellence do not come that easily.' I said, 'If you really want to learn how to do MOF chemistry, you better come and work with me.' I think that shocked him, but here he is."

How did Deng react to Yaghi's offer?

"Definitely," said Deng, who plans to become a chemistry professor. "And," he added, "the story ends with me trying enough times to get it right. It took me about a hundred more times."

"With MOF chemistry," Yaghi said, "it is not all design; there is a lot of trial and error because we are trying to learn what nature is telling us, and learning that code takes time."

"What is special about DJ and the other students who have worked in my laboratory is that no matter how much you raise the bar, they jump high enough to rise above it," Yaghi said. "It takes a special student to do that, but they are out there, and they need to be inspired. Working with students like DJ that I can challenge in this way is every professor's dream."

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